

States of Matter and Phase Changes

Matter can exist in three major states:

→ **Solid** → Low temperature and low energy.

Has a definite volume and a definite shape.

Draw each state below:



→ **Liquid** → Takes the shape of its container.

Has a definite volume, but no definite composition.



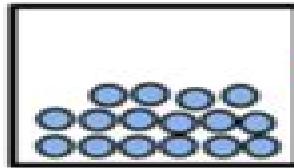
→ **Gas** → High amount of energy and disorder.

Has no definite shape and no definite composition.



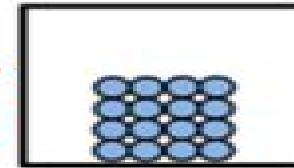
Instructions: Identify the phase changes below and whether energy is increasing or decreasing.

1)



Energy

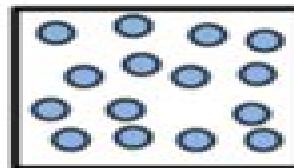
Increase **or** decrease



Phase Change:

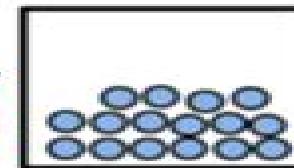
Freezing

2)



Energy

Increase **or** decrease



Phase Change:

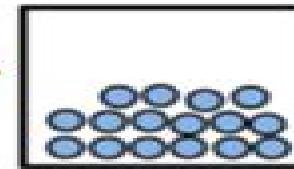
Condensation

3)



Energy

Increase **or** decrease



Phase Change:

Melting

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