

# States of Matter and Phase Changes

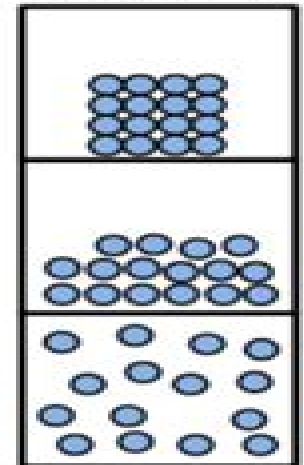
Matter can exist in three major states:

➔ **Solid** → Low temperature and low energy.  
Has a definite volume and a definite shape.

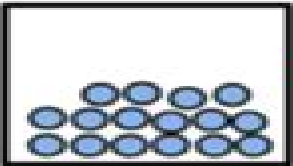
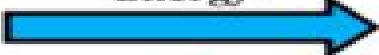
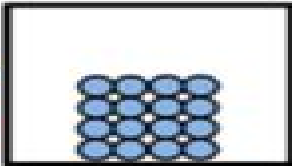
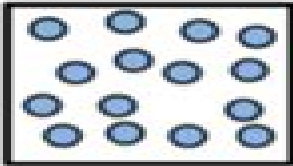

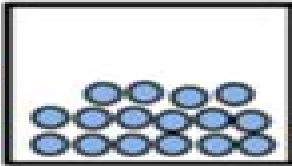
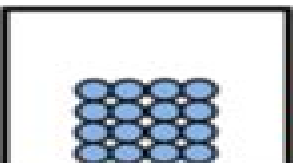

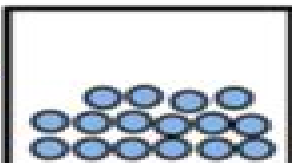
➔ **Liquid** → Takes the shape of its container.  
Has a definite volume, but no definite composition.

➔ **Gas** → High amount of energy and disorder.  
Has no definite shape and no definite composition.

Draw each state below:



Instructions: Identify the phase changes below and whether energy is increasing or decreasing.

1)		 Increase <del>or</del> decrease		Phase Change: <b>Freezing</b>
2)		 Increase <del>or</del> decrease		Phase Change: <b>Condensation</b>
3)		 Increase <del>or</del> decrease		Phase Change: <b>Melting</b>

# Phases Of Matter Guided Answers

**Arihant Experts**



## **Phases Of Matter Guided Answers:**

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