

GAS LAW WORKSHEET 1 KEY

1. A sample of oxygen gas occupies a volume of 436.2 mL at 1.02 atm. If the temperature is held constant, what would the pressure of this gas be when the gas is compressed to 231.6 mL?

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \text{since } T_1 = T_2 \quad P_1 V_1 = P_2 V_2 \quad P_2 = P_1 \frac{V_1}{V_2} = 1.02 \text{ atm} \left(\frac{436.2 \text{ mL}}{231.6 \text{ mL}} \right) = 1.92 \text{ atm}$$

To check with reasoning: decreased V leads to increased P, so the volume ratio should be greater than one to yield the expected increased pressure.

2. If a gas originally occupying 6.75 L at 19.21 °C and 762.5 torr is compressed to give a pressure of 1.26 atm at 26.35 °C, what would the new volume be?

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad V_2 = V_1 \left(\frac{T_2}{T_1} \right) \left(\frac{P_1}{P_2} \right) = 6.75 \text{ L} \left(\frac{299.50 \text{ K}}{292.36 \text{ K}} \right) \left(\frac{762.5 \text{ torr}}{1.26 \text{ atm}} \right) \left(\frac{1 \text{ atm}}{760 \text{ torr}} \right) = 5.51 \text{ L}$$

To check with reasoning: increased T leads to increased V, so the temperature ratio should be greater than one to yield the expected increase in volume. Increased P leads to decreased V, so the pressure ratio should be less than one to yield the expected decreased volume.

3. Calculate the number of grams of hydrogen sulfide gas, H₂S, in a 3.6 L container at 32.6 °C and 712 mmHg.

$$PV = \frac{g}{M} RT \quad g = \frac{PVM}{RT} = \frac{712 \text{ mmHg} (3.6 \text{ L}) 34.080 \frac{\text{g}}{\text{mol}}}{0.082058 \frac{\text{L atm}}{\text{K mol}} (305.8 \text{ K})} \left(\frac{1 \text{ atm}}{760 \text{ mmHg}} \right) = 4.6 \text{ g H}_2\text{S}$$

4. Calculate the density of 0.625 g of carbon dioxide at 26.32 °C and 1.03 atm.

$$PV = \frac{g}{M} RT \quad \frac{g}{V} = \frac{PM}{RT} = \frac{1.03 \text{ atm} \left(44.010 \frac{\text{g}}{\text{mol}} \right)}{0.082058 \frac{\text{L atm}}{\text{K mol}} (299.47 \text{ K})} = 1.84 \text{ g/L CO}_2$$

Phet Idael Gas Law Answer Key

Matthew Schnurr, Anna MacLeod

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A Guide to Teaching in the Active Learning Classroom Paul Baepler,J. D. Walker,D. Christopher Brooks,Kem Saichaiie,Christina I. Petersen,2023-07-03 While Active Learning Classrooms or ALCs offer rich new environments for learning they present many new challenges to faculty because among other things they eliminate the room's central focal point and disrupt the conventional seating plan to which faculty and students have become accustomed The importance of learning how to use these classrooms well and to capitalize on their special features is paramount The potential they represent can be realized only when they facilitate improved learning outcomes and engage students in the learning process in a manner different from traditional classrooms and lecture halls This book provides an introduction to ALCs briefly covering their history and then synthesizing the research on these spaces to provide faculty with empirically based practical guidance on how to use these unfamiliar spaces effectively Among the questions this book addresses are How can instructors mitigate the apparent lack of a central focal point in the space What types of learning activities work well in the ALCs and take advantage of the affordances of the room How can teachers address familiar classroom management challenges in these unfamiliar spaces If assessment and rapid feedback are critical in active learning how do they work in a room filled with circular tables and no central focus point How do instructors balance group learning with the needs of the larger class How can students be held accountable when many will necessarily have their backs facing the instructor How can instructors evaluate the effectiveness of their teaching in these spaces This book is intended for faculty preparing to teach in or already working in this new classroom environment for administrators planning to create ALCs or experimenting with provisionally designed rooms and for faculty developers helping teachers transition to using these new spaces

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