

# A-Level Chemistry - Year 2

## OCR Chemistry A - Paper 1 June - 2019

16 Sir Humphry Davy discovered several elements including sodium, potassium, magnesium, calcium and chromium.

(a) Explain which block in the Periodic Table sodium and magnesium belong to.

Question - 16

[1]

(b) A sample of magnesium,  $A_e = 24.305$ , is found to consist of three isotopes. The accurate relative isotopic masses and % abundances of two of the isotopes are shown in the table.

Isotope	Relative isotopic mass	% abundance
$^{24}\text{Mg}$	23.985	78.99%
$^{26}\text{Mg}$	24.986	10.00%

Determine the relative isotopic mass of the third isotope of magnesium in the sample.

Give your answer to 3 significant figures.

relative isotopic mass = \_\_\_\_\_ [1]

OCR

International and

Tuesday 4 June 2019 - Afternoon

A Level Chemistry A

H430/01 Periodic table, elements and physical chemistry

Two hours 1 hour 15 minutes

Specimen Test

1 mark and no diagrams  
and no working shown

Working out

1 mark with a correct answer



Please write clearly in block capitals (do not write in the margins).

Centre number

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Last name

Answers/Working

- 1 mark each pt. You may use an A4 sheet for graphs and diagrams.
- Answers of the questions.
- Where appropriate, your working should be indicated with working marks may be given for a correct method even if the answer is incorrect.
- Where your answer to each question is in excess of one page, if further space is required, use the back pages at the end of this paper. The question number must be clearly shown.

Information

- The total mark for this paper is 100.
- The marks for each question are shown in brackets if 0.
- Quality of extended responses will be assessed in questions marked with an asterisk (\*).
- This document consists of 10 pages.

# Ocr Chemistry June 2013 Past Paper F321

**Sabine Zange**

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## **Table of Contents Ocr Chemistry June 2013 Past Paper F321**

1. Understanding the eBook Ocr Chemistry June 2013 Past Paper F321
  - The Rise of Digital Reading Ocr Chemistry June 2013 Past Paper F321
  - Advantages of eBooks Over Traditional Books
2. Identifying Ocr Chemistry June 2013 Past Paper F321
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Ocr Chemistry June 2013 Past Paper F321
  - User-Friendly Interface
4. Exploring eBook Recommendations from Ocr Chemistry June 2013 Past Paper F321
  - Personalized Recommendations
  - Ocr Chemistry June 2013 Past Paper F321 User Reviews and Ratings
  - Ocr Chemistry June 2013 Past Paper F321 and Bestseller Lists
5. Accessing Ocr Chemistry June 2013 Past Paper F321 Free and Paid eBooks
  - Ocr Chemistry June 2013 Past Paper F321 Public Domain eBooks
  - Ocr Chemistry June 2013 Past Paper F321 eBook Subscription Services
  - Ocr Chemistry June 2013 Past Paper F321 Budget-Friendly Options
6. Navigating Ocr Chemistry June 2013 Past Paper F321 eBook Formats

- ePub, PDF, MOBI, and More
- Ocr Chemistry June 2013 Past Paper F321 Compatibility with Devices
- Ocr Chemistry June 2013 Past Paper F321 Enhanced eBook Features

## 7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Ocr Chemistry June 2013 Past Paper F321
- Highlighting and Note-Taking Ocr Chemistry June 2013 Past Paper F321
- Interactive Elements Ocr Chemistry June 2013 Past Paper F321

## 8. Staying Engaged with Ocr Chemistry June 2013 Past Paper F321

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Ocr Chemistry June 2013 Past Paper F321

## 9. Balancing eBooks and Physical Books Ocr Chemistry June 2013 Past Paper F321

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Ocr Chemistry June 2013 Past Paper F321

## 10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

## 11. Cultivating a Reading Routine Ocr Chemistry June 2013 Past Paper F321

- Setting Reading Goals Ocr Chemistry June 2013 Past Paper F321
- Carving Out Dedicated Reading Time

## 12. Sourcing Reliable Information of Ocr Chemistry June 2013 Past Paper F321

- Fact-Checking eBook Content of Ocr Chemistry June 2013 Past Paper F321
- Distinguishing Credible Sources

## 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

## 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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