

# PERIODIC PATTERNS

## READING COMPREHENSION WORKSHEET

<p>10. Indicate using electron shells are likely present.</p> <p>a. Yes b. No</p> <p>Write the correct word in each blank space.</p> <p>11. Elements in the same _____ have similar properties.</p> <p>12. As atomic number increases across a period, atomic radius _____.</p> <p>13. The electron arrangement is identical in all elements in the same group.</p> <p>14. High _____ energy means electrons are further from the nucleus.</p> <p>15. Elements in the same group have similar chemical properties.</p> <p>16. As atomic number increases across a period, atomic radius _____.</p> <p>17. Elements in the same group have similar chemical properties.</p> <p>18. As atomic number increases across a period, atomic radius _____.</p> <p>19. Elements in the same group have similar chemical properties.</p> <p>20. As atomic number increases across a period, atomic radius _____.</p> <p>21. Elements in the same group have similar chemical properties.</p> <p>22. 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The periodic table is a powerful tool for understanding the relationships between elements and their properties.</p> <p>Elements in the same group have similar chemical properties, which is due to the similar arrangement of their outermost electrons. For example, elements in Group 1 (alkali metals) are highly reactive and form similar compounds, while elements in Group 18 (noble gases) are inert and do not react with other elements. The periodic table also shows trends in atomic size, with atoms becoming larger as you move down the table and smaller as you move across the table.</p> <p>Elements in the same period have the same number of electron shells, but their chemical properties vary. For example, elements in Period 2 range from highly reactive metals (like lithium) to non-reactive gases (like neon). The periodic table is a powerful tool for understanding the relationships between elements and their properties.</p> <p>Read the passage carefully before answering the questions that follow.</p> <p>1. Which of the following best describes the organization of elements in the periodic table?</p> <p>a. Elements are arranged by increasing atomic number. b. Elements are arranged by increasing atomic mass. c. Elements are arranged by increasing electronegativity. d. Elements are arranged by increasing atomic size.</p> <p>2. Which of the following best describes the relationship between elements in the same group?</p> <p>a. They have similar chemical properties. b. They have similar physical properties. c. They have similar atomic masses. d. They have similar electronegativities.</p> <p>3. Which of the following best describes the relationship between elements in the same period?</p> <p>a. They have the same number of electron shells. b. They have the same chemical properties. c. 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They have similar physical properties. c. They have similar atomic masses. d. They have similar electronegativities.</p>
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MS. EMILY HAYES

NGSS - GRADE 11 - SCIENCE - ANSWER KEY

# Periodic Patterns Worksheet Answer Key

**Mike Girvin**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## Periodic Patterns Worksheet Answer Key:

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket The IT in Secondary Science Book Roger Frost, 1994 **Science Spectrum** Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2003-03 *Cambridge IGCSETM Chemistry Teacher's Guide (Collins Cambridge IGCSETM)* Chris Sunley, 2022-02-03 Prepare students with complete coverage of the revised Cambridge IGCSETM Chemistry syllabus 0620 0971 for examination from 2023 Collins Cambridge IGCSE Chemistry Teacher s Guide is full of lesson ideas practical instructions technician s notes planning support and more **Educart ICSE Class 10 One-shot Question Bank 2026 Chemistry (strictly for 2025-26 boards)** Sir Tarun Rupani, 2025-07-12 Fast track your Chemistry revision with this exam ready resource This One shot Question Bank by Sir Tarun Rupani is designed to help ICSE Class 10 students revise the complete Chemistry syllabus quickly and thoroughly It simplifies theory boosts numerical accuracy and ensures strong exam practice all aligned with the 2025 26 ICSE syllabus Key Features Strictly Based on ICSE 2025 26 Curriculum Complete chapter coverage including Periodic Table Chemical Bonding Acid Base Organic Chemistry and more One shot Format Each chapter includes concise concept notes chemical equations reactions and key diagrams for quick recall Complete Coverage of Question Types Includes objective short long answers equation based numerical and reasoning questions Chapterwise PYQs Included Practice with previous years ICSE board questions to understand trends and improve retention Solved Answers in ICSE Format Clear well structured solutions using proper units chemical symbols and balanced equations Smart Revision Focus Special tips to avoid common mistakes in writing reactions balancing equations and attempting numericals Why Choose This Book This Chemistry One shot by Sir Tarun Rupani is built for smart preparation whether you re revising at the last minute or practising throughout the term It helps you approach each question with clarity confidence and the precision needed to score high in the 2026 ICSE board exam Microsoft 365 Excel: The Only App That Matters Mike Girvin, 2021-04-01 This is a book about Microsoft 365 Excel or Excel 365 No other version in history has as much power as many features or more possibilities than does Excel 365 With a new formula calculations engine and many new built in functions creating formula solutions and business models in Excel 365 is dramatically easier than at any time in the history of spreadsheets In addition with the new data tools like Power Query Power Pivot and Power BI performing data analysis to make data driven decisions can be easily

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