

2.6 - Periodic Trends worksheet

Name: _____

I. Short answer

1. Label each group on the periodic table.
2. Explain what valence electrons are.
3. Explain what makes an element stable?

II. Answer the questions using the symbols on the periodic table.

1. What elements are in period 5?

What element(s) is an

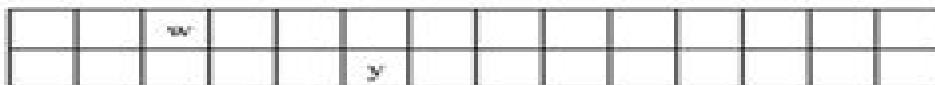
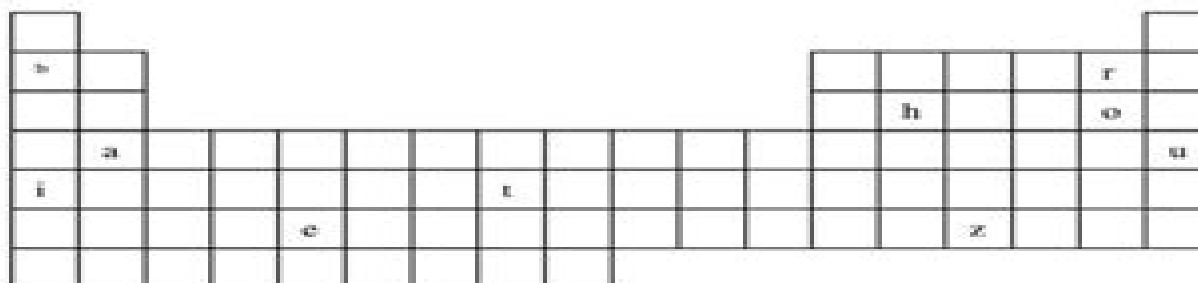
| | |
|--------------------------|---------------------|
| 2. halogens | 6 transition metals |
| 3. lanthanides | 7 alkali metals |
| 4. alkaline earth metals | 8 actinides |
| 5. noble gases | |

What elements are in the sublevel

— 9 —
— 10 —
— 11 —
— 12 —

What elements are

13. metals,
14. nonmetals
15. metalloids,
16. Which element is the most reactive nonmetal?
17. Which element is the most reactive metal?
18. Which element below is the least reactive?
19. Which element has the largest radius in period 3?
20. Which element has the largest radius in the Alkali metals?
21. Which element has the highest ionization energy?
22. Which elements has the second largest atomic radius?
23. Which element has the largest atomic mass?
24. Which element is located at period 4, group 15?
25. Which element is located at period 5, group 2?
26. Which elements will form +1 ions?
27. Which group of metals will have the same properties?
28. Which group of nonmetals will have the same properties?
29. On the back, define all chapter 5, section 3 vocabulary words.
30. On the back, draw the periodic table.



Periodic Table Chemistry Answers

Bernice G. Segal, Peter S. Shenkin

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