

PERIODIC TRENDS- CHEMISTRY FINAL EXAM GUIDE

2025

| MOST COMMON QUESTIONS WITH CORRECTLY VERIFIED ANSWERS | ALREADY A+ GRADED | GUARANTEED PASS

As you move from top to bottom down a group on the periodic table the size of an atom will _____ - ✓ ✓ increase

This happens because as you move down the group each element has one more occupied _____ than the last one. - ✓ ✓ energy level

As you move from left to right across a period on the periodic table the size of an atom will _____ - ✓ ✓ decrease

This happens because as you move across a period each element has one more _____ in its nucleus than the last one. - ✓ ✓ proton

Within a period the elements will have the same number of occupied _____ - ✓ ✓ energy levels

The _____ is caused by inner electrons interfering with the attraction of outer electrons for the protons in the nucleus. - ✓ ✓ shielding effect

Electrons which are further from the nucleus are _____ attracted to the nucleus because of this effect. - ✓ ✓ less

The decrease in atomic size as you move from left to right across the periodic table becomes less pronounced as you move _____ the periodic table. - ✓ ✓ down

Explain why fluorine has a smaller atomic radius than oxygen. - ✓ ✓ Due to fluorine having one more proton in its nucleus. (More nuclear charge pulling the energy levels in closer.)

Explain why fluorine has a smaller atomic radius than chlorine. - ✓ ✓ Due to chlorine having an extra occupied energy level

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M.D., 2013-03-05 Offers test strategies reviews key concepts of chemistry and provides three full length practice tests with answers and explanations **Cracking the SAT Chemistry Subject Test, 15th Edition** Princeton Review, 2015-02-17 EVERYTHING YOU NEED TO HELP SCORE A PERFECT 800 Equip yourself to ace the SAT Chemistry Subject Test with The Princeton Review's comprehensive study guide including 3 full length practice tests thorough reviews of key chemistry topics and targeted strategies for every question type This eBook edition has been optimized for on screen viewing with cross linked questions answers and explanations We don't have to tell you how tough SAT Chemistry is or how helpful a stellar exam score can be for your chances of getting into your top choice college Written by the experts at The Princeton Review *Cracking the SAT Chemistry Subject Test* arms you to take on the test and achieve your highest score Techniques That Actually Work Tried and true strategies to help you avoid traps and beat the test Tips for pacing yourself and guessing logically Essential tactics to help you work smarter not harder Everything You Need to Know to Help Achieve a High Score Expert subject reviews for every test topic Up to date information on the SAT Chemistry Subject Test Score conversion tables for accurate self assessment Practice Your Way to Perfection 3 full length practice tests with detailed answer explanations Hands on experience with all three question types in each content chapter Complete study sheet of core formulas and terms

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Chemistry in a Month John Wilkinson, 2003 *Health Trends in Secondary Education* American Child Health Association, 1927 *How to Make the Periodic Health Examination* Eugene Lyman Fisk, J. Ramser Crawford, 1927

Journal of the Pennsylvania Academy of Science, 2001 *McGraw-Hill Education SAT Subject Test Chemistry, Fifth Edition* Thomas A. Evangelist, 2018-11-23 We Will Help You Get Your Best Score With more than 125 years of experience in

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McGraw-Hill Education SAT Subject Test Chemistry 4th Ed. Thomas A. Evangelist, 2016-01-01
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Chemistry 2007 David Wilson, 2006-12 Kaplan's guide includes 2 full length practice tests Diagnostic test to target areas for score improvement Detailed answer explanations Hundreds of practice questions from calculations of chemical equations to organic chemistry Explanations of important terms formulas and concepts Powerful strategies to help you score higher

Carolina Science and Math Carolina Biological Supply Company, 2003

California's Health, 1946

A Case Study of Prior Knowledge, Learning Approach and Conceptual Change in an Introductory College Chemistry Tutorial Program Per Christian Braathen, 1987

Metals Handbook: Mechanical testing, 1978

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