

## CHEMISTRY 9TH - Multiple Choice Questions

# → PERIODIC TABLE AND PERIODICITY OF PROPERTIES

CHAPTER# 03

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01. He proposed the classification of Triads in which several groups of three elements classified based on atomic masses.
- (a) ✓ Dobereiner (b) Newland  
(c) Mendeleev (d) Moseley
02. In the classification of Triads central element had an atomic mass average of the other:
- (a) Five elements (b) Four elements  
(c) Three elements (d) ✓ Two elements
03. In 1864 British chemist Newland put forward the:
- (a) classification of Triads (b) ✓ Law of Octaves  
(c) 8 columns and rows arrangement (d) 9 columns and groups arrangement
04. It is order of increasing atomic masses.
- (a) classification of Triads (b) ✓ Law of Octaves  
(c) 8 columns and rows arrangement (d) 9 columns and groups arrangement
05. According to him the eighth element has similar properties as the first element in a group of eight elements.
- (a) Dobereiner (b) Moseley  
(c) ✓ Newland (d) Mendeleev
06. In 1869, he published eight vertical columns (groups) and horizontal rows (periods) on the basis of physical and chemical properties of elements.
- (a) Dobereiner (b) Newland  
(c) ✓ Mendeleev (d) Moseley

# Periodicity Multiple Choice Questions

**John Moore, Richard H. Langley**



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