

CHEMISTRY 9TH – Multiple Choice Questions

→ PERIODIC TABLE AND PERIODICITY OF PROPERTIES

CHAPTERS 03

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01. He proposed the classification of Triads in which several groups of three elements classified based on atomic masses.

(a) Dobereiner (b) Newland
(c) Mendeleev (d) Moseley

02. In the classification of Triads central element had an atomic mass average of the other:

(a) Five elements (b) Four elements
(c) Three elements (d) Two elements

03. In 1864 British chemist Newland put forward the:

(a) classification of Triads (b) Law of Octaves
(c) 8 columns and rows arrangement (d) 9 columns and groups arrangement

04. It is order of increasing atomic masses.

(a) classification of Triads (b) Law of Octaves
(c) 8 columns and rows arrangement (d) 9 columns and groups arrangement

05. According to him the eighth element has similar properties as the first element in a group of eight elements.

(a) Dobereiner (b) Moseley
(c) Newland (d) Mendeleev

06. In 1869, he published eight vertical columns (groups) and horizontal rows (periods) on the basis of physical and chemical properties of elements.

(a) Dobereiner (b) Newland
(c) Mendeleev (d) Moseley

Periodicity Multiple Choice Questions

John Moore, Richard H. Langley

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