

Chemistry I: Periodicity Worksheet

- 1) Arrange the following elements in order of increasing size: Al, Br, C, K, and Na.

100

- 2) Arrange the following elements in order of increasing size: Ca, Rb, P, Ge, and Sr.

10

- 3) Circle the atom or ion in each pair that has the larger radius.

- ch In or I

- 4) Select the atom or ion in each pair that has the larger radius.

- ch Error As

- 5) Which of the following groups of elements is arranged correctly in order of increasing ionization energy?
(Answer by circling the correct answer.)

40. $\text{Na} < \text{C} < \text{Si} < \text{Li}$

- 6) Arrange the following atoms in order of increasing ionization energy: Li, K, C, and N.

- 7) Arrange the following atoms in order of increasing ionization energy: Si, K, As, and Ca.

- 8) Compare the elements Na, Mg, O, and P.

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Periodicity Worksheet Answer Key

Ernest Sydney Hedges



Periodicity Worksheet Answer Key:

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