

## Periodic Trends

**ATOMIC RADIUS**

1. What trend in atomic radius do you see as you go down a group/family on the periodic table?
2. What causes this trend?
  
3. What trend in atomic radius do you see as you go across a period/row on the periodic table?
4. What causes this trend?
  
5. Circle the atom in each pair that has the largest atomic radius.  
a) Al B      b) S O      c) Br Cl  
d) Na Al      e) O F      f) Mg Ca

6. Put the following elements in order from smallest to largest atomic radius **and explain why**:  
C, O, Sn, Sr.

**ELECTRONEGATIVITY**

7. Define electronegativity
  
8. How does the ionic radius of a nonmetal compare with its atomic radius?
  
9. What trend in electronegativity do you see as you go down a group/family on the periodic table?
  
10. What causes this trend?
  
11. What trend in electronegativity do you see as you go across a period/row on the periodic table?
  
12. What causes this trend?
  
  
13. Circle the atom in each pair that has the greater electronegativity.  
a) Ca Ga      b) Li O      c) Cl S      d) Br As      e) Ba Sr      f) O S

**GENERAL QUESTIONS**

14. Which group tends to form +1 ions? \_\_\_\_\_
15. Which group tends to form +2 ions? \_\_\_\_\_
16. Which group tends to form -1 ions? \_\_\_\_\_
17. Which group tends not to form ions or react? \_\_\_\_\_

# Periodic Table Trends Activity Answer Key

**American Chemical Society**

## **Periodic Table Trends Activity Answer Key:**

Hands-On General Science Activities With Real-Life Applications Pam Walker, Elaine Wood, 2008-04-21 In this second edition of Hands On General Science Activities with Real Life Applications Pam Walker and Elaine Wood have completely revised and updated their must have resource for science teachers of grades 5-12. The book offers a dynamic collection of classroom ready lessons, projects and lab activities that encourage students to integrate basic science concepts and skills into everyday life.

**ChemDiscovery Student Guide** Olga I. Agapova, 2002

**Problems and Problem Solving in Chemistry Education** Georgios Tsaparlis, 2021-05-17

Problem solving is central to the teaching and learning of chemistry at secondary, tertiary and post tertiary levels of education, opening to students and professional chemists alike a whole new world for analysing data, looking for patterns and making deductions. As an important higher order thinking skill, problem solving also constitutes a major research field in science education. Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving. The following situations are considered: some general others with a focus on specific areas of chemistry: quantitative problems, qualitative reasoning, metacognition and resource activation; deconstructing the problem solving process; an overview of the working memory, hypothesis reasoning with the electron pushing formalism, scaffolding organic synthesis skills, spectroscopy for structural characterization in organic chemistry, enzyme kinetics, problem solving in the academic chemistry laboratory, chemistry problem solving in context, team based active learning, technology for molecular representations, IR spectra simulation and computational quantum chemistry tools. The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry. With a foreword by George Bodner.

*Chemistry in the Community (ChemCom)* American Chemical Society, 2011-06-17

Touted as the most successful NSF funded project published, Chemistry in the Community, ChemCom by the American Chemical Society, ACS, offers a meaningful and memorable chemistry program for all levels of high school students. ChemCom covers traditional chemistry topics within the context of societal issues and real world scenarios. Centered on decision making activities where students are responsible for generating data in an investigating, analyzing that data and then applying their chemistry knowledge to solve the presented problem. The text is intensively laboratory based with all 39 of the investigations integrated within the text, not separate from the reading. With the ChemCom program, students learn more organic and biochemistry, more environmental and industrial chemistry and more on the particulate nature of matter than other textbooks. All within the relevance of solving problems that arise in everyday life. Meticulously updated to meet the needs of today's teachers and students, the new sixth edition of ChemCom adheres to the new science framework as well as the forthcoming next generation of science standards. Incorporating advances in learning and cognitive sciences, ChemCom's wide ranging coverage builds upon the concepts and principles found in the National Science Education Standards. Correlations are

available showing how closely aligned ChemCom is to these and other state standards      **Learning and Leading with Technology** ,2001    **Edexcel A Level Chemistry Student Book 1** Andrew Hunt,Graham Curtis,Graham Hill,2015-07-17 Exam Board Edexcel Level AS A level Subject Chemistry First Teaching September 2015 First Exam June 2016 Endorsed by Edexcel Develop and assess your students knowledge and mathematical skills throughout A Level with worked examples practical assessment guidance and differentiated end of topic questions with this Edexcel Year 1 student book Identifies the level of your students understanding with diagnostic questions and a summary of prior knowledge at the start of the Year 1 Student Book Provides support for all 16 required practicals with various activities and questions along with a Practical chapter covering procedural understanding and key ideas related to measurement Mathematical skills are integrated throughout with plenty of worked examples including notes on methods to help explain the strategies for solving each type of problem Offers plenty of practice with Test Yourself Questions to help students assess their understanding and measure progress Encourages further reading and study with short passages of extension material Develops understanding with free online access to Test yourself Answers and an Extended Glossary Edexcel A level Chemistry Year 1 Student Book includes AS level    Prentice Hall Chemistry ,2000    Matter, Building Block of the Universe ,1993    **Princeton Review AP**

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*Chemistry Teacher* ,2001    **Atoms, Molecules & Elements: Patterns In the Periodic Table Gr. 5-8** George Graybill,2015-10-01 This is the chapter slice Patterns In the Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom's Taxonomy and STEM initiatives    *Hospital Administration Handbook* Howard S.

Rowland, Beatrice L. Rowland, 1984

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