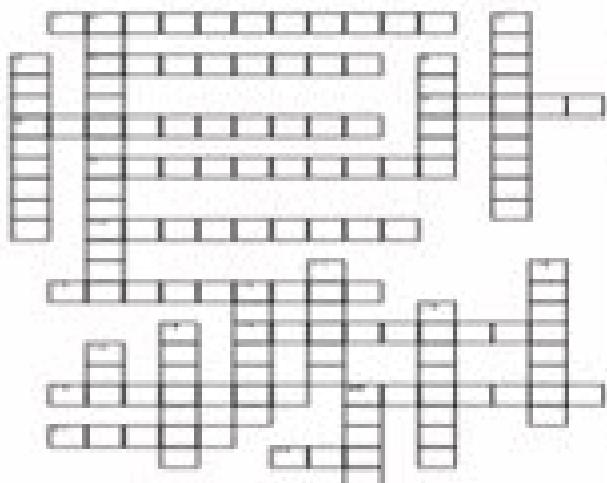


Periodic Table Assignment

Read pages 14 – 19. Complete the crossword puzzle (in pencil) and fill in the periodic table as instructed below.



ACROSS

1. Synthetic elements with atomic numbers greater than 92
5. This law states that the properties of elements repeat as a function of their atomic number
(10 letters in this phrase)
8. Elements were originally placed in order of atomic _____, now they are in order of atomic _____
9. The group that consists of very un-reactive gases
10. Number of elements known in 1860
13. Element 101 was named in his honour
17. The periodic table is normally drawn with 18 columns. A more accurate representation has _____ columns
19. Most elements are in this state
20. Element that is a liquid at SATP
21. The general name given to a column in the periodic table
22. According to Figure 6, when aluminum combines with oxygen, three oxygen atoms combine with this many aluminum atoms

DOWN

2. Elements that most closely follow the periodic law
3. The first scientist to organize elements based on repeating patterns
6. The modern name for aluminum
7. Most elements are _____, since they fall to the left of the "staircase line"
11. The general name given to a row (left-to-right) in the periodic table
12. Iodine belongs to this group
14. The first alkali metal
15. This element was assigned a relative atomic mass of one because it is the lightest element
16. The state of matter represented by (loop)
18. Silicon is in this group
20. The staircase that separates metals and non-metals begins below this element

1	2	3
4	5	6
7	8	9
10	11	12
13	14	15
16	17	18
19	20	21
22	23	24

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24
25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48
49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72
73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24
25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48
49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72
73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96

- 1) Place a large G in boxes that house elements that are gases at SATP, and a large L in boxes that contain elements that are liquids. The rest of the elements are solids, draw wavy lines in these boxes.
- 2) Draw the "staircase line" that divides metals from non-metals.
- 3) Semimetals (i.e. semi-conductors) are elements that have properties midway between metals and non-metals. The metalloids are: Si, Si, Ge, As, Sb, Te, Po, At. Identify the metalloids by drawing diagonal lines in these boxes.
- 4) Notice that each group (column) in the periodic table is identified on the top with a number and/or letter. There are three conventions used in the textbook. Label the groups that end in A (e.g. IA, IIIA, etc.).
- 5) Draw and underline the following groups (most II is not an IA metal): Alkali metals, alkaline earth metals, halogens, noble gases, transition metals, the inner transition elements (made up of lanthanides and actinides).
- 6) Create a legend for your periodic table.

Periodic Table Assignment Answer Key

Prentice-Hall Staff

Periodic Table Assignment Answer Key:

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learning materials effectiveness problem solving and even students approaches during experimentation Eye movements as captured using eye tracking can reveal information about a student's attention and cognition on a process level going well beyond classical product based assessment techniques such as questionnaires or tests THE Journal ,1999

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