

Periodic Table of Elements

Reading Comprehension Passage

Periodic Table of Elements

Name: _____

Periodic Table Passage

- _____ is the element that has the most protons.
- _____ is the element that has the most neutrons.
- _____ is the element that has the most electrons.
- _____ is the element that has the most protons and neutrons.
- _____ is the element that has the most neutrons and electrons.
- _____ is the element that has the most protons and electrons.
- _____ is the element that has the most neutrons and protons.
- _____ is the element that has the most protons, neutrons, and electrons.
- _____ is the element that has the most neutrons, protons, and electrons.
- _____ is the element that has the most protons, neutrons, and electrons.

Periodic Table of Elements

Name: _____

Periodic Table Passage

Periodic Table Terms

- The _____ lists elements in order of increasing atomic mass.
- Each element's entry includes its name, symbol, atomic mass, and _____.
- The elements are arranged in vertical columns called _____.
- Horizontal rows are called _____.
- Elements in the same group have similar properties. For example, all the elements in Group 1 are very reactive, while the elements in Group 2 are very stable.

Periodic Groups

- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____
- _____

Periodic Table of Elements

The Periodic Table of Elements is like a big puzzle that helps scientists understand how all the different elements in the world fit together. It's kind of like a big family tree for all the elements!

The Periodic Table has all the elements listed in order from the lightest to the heaviest. Each element has its own special box on the table with its own unique symbol and number. The number tells us how many protons are in the nucleus of the atom of that element.

There are over 100 different elements on the Periodic Table! Some elements you might have heard of are oxygen, gold, and carbon. Each element has its own special properties, like how it looks, how it behaves, and what it can be used for.

The Periodic Table is divided into rows and columns. The rows are called periods, and the columns are called groups. Elements in the same group have similar properties. For example, all the elements in Group 1 are very reactive, while the elements in Group 2 are very stable.

The Periodic Table was first created by a scientist named Dmitri Mendeleev in 1869. He noticed that when he arranged the elements in order of their atomic mass, certain patterns started to emerge. He was able to predict the properties of elements that hadn't even been discovered yet!

Thanks to Mendeleev's work, we now have a handy tool that helps us understand the building blocks of the universe. Scientists use the Periodic Table to study how elements combine to form compounds, and how they react with each other.

So the next time you see the Periodic Table, remember that it's like a big family tree for all the elements in the world. Each element has its own unique place on the table, and together they make up everything around us!

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